



ENERGY WAVE of the Periodic Table Definitive Periodic Law REVEALED

William C. Scott

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Definitive Periodic Law is revealed in arrangement of new Periodic Table, repeating sequential numbers of protons discovered in Groups 1 through 18 in the elements of the new ENERGY WAVE of the Periodic Table. The elements in the new ENERGY WAVE of the Periodic Table are given in the ground state, which is one electron for each proton. This arrangement provided a unique opportunity to observe the nucleus of the elements. By incorporating the sequential numbers of protons underlying the Energy Levels K, L, M, N, O, P, & Q in shell blocks s, p, d, and f of the Group elements, it revealed what had been hidden and veiled in the complexity of electron configurations. Sequential numbers of protons are observed to repeat in the Group elements from period to period. This is the true revealed energy force creating the similar physical and chemical properties of Groups 1 through 18 from period to period in the Periodic Table. The ENERGY WAVE of the Periodic Table had revealed Definitive Periodic Law... "Definitive Periodic Law is the number of protons underlying the Energy Levels K, L, M, N, O, P, & Q in the nucleus of the Elements. These sequential numbers of protons repeat in shell blocks s, p, d, and f, forming groups that have similar physical and chemical properties from period to period." These sequential numbers of protons are the cornerstones of the nucleus and provide the atomic orbitals of the electrons the foundation for their spatial relationship to the nucleus as described by the azimuthal (angular) and magnetic numbers of quantum chemistry. These sequential numbers of protons are very important, as they reveal new explanation to chemical bond angles and the molecular geometry and structure of molecules.

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